

(12) UK Patent Application (19) GB (11) 2 335 428 (13) A

(43) Date of A Publication 22.09.1999

(21) Application No 9805880.3	(51) INT CL ⁶ E21B 33/138 , C08F 8/28 220/56 // (C08F 220/56 220:18)
(22) Date of Filing 20.03.1998	(52) UK CL (Edition Q) C3P PFE P220 P250 P268 P310 P318 P320 P324 C3J JCC C3W W210 E1F FPC
(71) Applicant(s) Sofitech N.V. (Incorporated in Belgium) Rue de Stalle 140, B-1180 Brussels, Belgium	(56) Documents Cited EP 0399767 A2 EP 0311799 A1 US 5003006 A US 4814096 A US 4520182 A US 4505827 A US 3984333 A
(72) Inventor(s) Timothy Gareth John Jones Gary John Tustin	(58) Field of Search UK CL (Edition P) C3J JCC , C3P PFE , E1F FPC INT CL ⁶ C08F 8/28 220/56 , E21B 33/138 43/22 ONLINE: WPI
(74) Agent and/or Address for Service Akram Karim Mirza Schlumberger Cambridge Research Limited, High Cross, Madingley Road, CAMBRIDGE, CB3 0EL, United Kingdom	

(54) Abstract Title
Hydrophobically modified polymers for water control

(57) A water soluble polymer comprises a substantially linear backbone along which hydrophobic functionalities are randomly located, which polymer is capable of being chemically cross-linked to produce a stable gel to block a water-bearing formation from a hydrocarbon-producing well. The polymer may have a molecular weight of 50,000 or more and preferably comprises 0.5 to 5 mole per cent of a hydrophobic monomer. The functional groups capable of being chemically cross-linked may be part of a hydrophilic or hydrophobic section of the polymer. A composition is further described which comprises the aforesaid polymer in combination with a chemical cross-linking agent. Typically, the cross-linking agent is an aldehyde or derivative thereof, having at least five carbon atoms, e.g. hexanal, heptanal.

GB 2 335 428 A

Hydrophobically Modified Polymers for Water Control

The present invention relates to cross-linkable hydrophobically modified water-soluble polymers for reducing the water

5 production of a hydrocarbon well drilled through subterranean formations. More particularly it relates to such polymers and polymer containing compositions for selectively reducing the flow of subterranean aqueous fluids into a well while maintaining the hydrocarbon production.

10

BACKGROUND OF THE INVENTION

During the life cycle of a hydrocarbon well, e.g., a well for
15 extracting oil or natural gas from the Earth, the producing well commonly also yields water. In these instances, the amount of water produced from the well tends to increase over time with a concomitant reduction of hydrocarbon production. Frequently, the production of water becomes so profuse that remedial measures
20 have to be taken to decrease the water/hydrocarbon production ratio. As a final consequence of the increasing water production, the well has to be abandoned.

Various techniques have been developed and used for reducing the
25 quantity of water produced from oil and/or gas wells. For example, US Pat. No. 3,719,228 discloses a method of treating a subterranean formation containing hydrocarbons and brine to stimulate the production of hydrocarbons: A preflush composition comprised of a water solution of rosin soap and fatty acid soap
30 is injected into the formation. The preflush reacts with connate brine to produce a precipitate that blocks the brine-bearing passages. The composition does not react with hydrocarbons thereby allowing hydrocarbon-bearing passages to remain open.

Another alternative method is disclosed in US Pat. No.

4,617,132: A sandstone formation is contacted with an aqueous solution containing a water soluble anionic polymer having a molecular weight greater than 100,000. As the next step, the

5 anionic polymer is contacted with a fluid containing a water soluble cationic polymer having a weight greater than 1,000. As a result of the contact of the anionic with the cationic polymer, coacervation occurs between the two polymers which reduces the amount of the anionic polymer removed from the
10 formation by fluids produced therefrom. The presence of stabilized polymer in the formation reduces the water/oil ratio by reducing the permeability of the formation to water in the wellbore area.

15 In yet another technique, disclosed in US Pat. No. 5,203,834, comprises the steps of injecting a gas, a polymer composition capable to form a foamed gel with the gas, and a delayed gel degrading agent being capable of opening pathways in the gel.

20 The gelation of high molecular weight polymers ($M_w > 10^6$ g/mol) has been extensively used in the development of water-based treatment fluids for water control, is further described for example by R. D. Sydansk in "Acrylamide-polymer/chromium(III)-carboxylate gels for near wellbore matrix treatments", 7th SPE
25 Symp. Enhanced Oil Recovery, Tulsa, Ok., April 1988, SPE/DoE 20214, or by R. S. Seright in: "Placement of gels to modify injection profiles", SPE/DoE Symp. Enhanced Oil Recovery, Tulsa, Ok., April 1994, SPE 27740. Typically for those methods, an aqueous solution of a high molecular weight polymer, such as a
30 polyacrylamide/polyacrylate copolymer (a so-called partially-hydrolysed polyacrylamide), is gelled *in situ* in a porous formation using a metal cross-linker such as Cr^{3+} or small water-soluble organic cross-linkers such as formaldehyde and

formaldehyde/phenol. Other water-soluble polymers such as poly(vinyl alcohol), the polysaccharide guar gum and the copolymer poly(vinylpyrrolidone-co-2-acrylamido-2-methyl-1-propanesulphonic acid) which can be cross-linked with a variety
5 of cross-linking agents such as Zr^{4+} and boric acid.

A more recent approach is described by A. Keller and K. A. Narh in: "The effect of counterions on the chain conformation of polyelectrolytes, as assessed by extensibility in elongational
10 flow: the influence of multiple valency", *J. Polym. Sci.: Part B: Polymer Phys.*, **32**, 1697-1706 (1994). It includes the cross-linking of poly(sodium 4-styrenesulphonate) using Al^{3+} ions to form a gel. The concentration of the high molecular weight hydrophilic polymers used to form hydrogels is typically in the
15 range 3-10 g/l.

Hydrophobically-modified polymers as such and their application as thickening agents in aqueous fluids has been extensively described. The principal application of these polymers is as
20 viscosifying agents and they are commonly referred to as associative thickeners; these polymers are also known as polysoaps. The associative behaviour is generated by a small mole fraction (typically ≤ 0.05) of hydrophobic groups which, as separate monomers, are not soluble in water. The hydrophobic
25 groups undergo intermolecular association in aqueous solution above the overlap concentration which results in a significant increase in solution viscosity.

Hydrophobically-modified polymers can be gelled in the presence
30 of monomeric surfactants. For example, large increases in the viscosity of solutions of hydrophobically-modified poly(acrylic acid) in the presence of anionic, cationic and non-ionic surfactants have been disclosed. A brief review of the

interaction between hydrophobically-modified polymers and surfactants can be found in: Lindman, B. and Thalberg, K., "Polymer-surfactant interactions - recent developments" in *Interactions of Surfactants with Polymers and Proteins*, E.D. Goddard and K.P. Ananthapadmanabhan (eds.), pp. 203-276, CRC Press, Boca Raton, Florida (1992).

In the U.S. Patent No. 4,532,052, the use of branched organic polymers for various well treating operations is described. The known polymers include hydrophilic and combinations of hydrophilic and hydrophobic side groups. The preferred examples are characterized as having large hydrophilic branches.

In the U.S. Patent No. 5,003,006, there are described cross-linked block copolymers containing polar and non-polar segments, with the polar segments generally making up at least 50 per cent.

It is the object of the invention to provide new polymers for water control application.

SUMMARY OF THE INVENTION

According to a first aspect of the invention, hydrophobically modified water-soluble polymers are provided which are capable of being chemically cross-linked to produce permanent gels in water-bearing formations. The polymers are characterized by a linear chain or backbone and attached thereto hydrophobic side groups at random locations. The term "linear" is defined as the absence of extended side groups/chains other than the hydrophobic groups. The term "extended" can be interpreted as having a molecular weight of more than 200.

It has been found that hydrophobically modified polymers when based on block copolymers, tend to form an inferior gel presumably due to presence of inhomogeneous gelation. In

5 addition there is economic advantage when producing polymers with high molecular weight through a random co-polymerization process. The use of large polymers is advantageous as it reduces the polymer concentration required to achieve a desired level of viscosity. Within the scope of the present invention, a high
10 molecular weight can be defined as a molecular weight in excess of 50,000, more preferably in excess of 100,000. A preferable upper limit for the size of the polymers is given by a molecular weight of 15 million, preferably 5 million, more preferably 500,000 or less.

15

The polymers preferably comprise low concentrations (0.5-5 mole percent) of hydrophobic monomers which randomly co-polymerise with the hydrophilic monomers to form random copolymers. The hydrophilic units of the polymer, which constitute preferably
20 95-99.5 mole percent of the polymer chain, contain preferably at least some small fraction of functional groups which can be chemically cross-linked to produce a gel. Alternatively, the hydrophobic groups themselves can contain chemical functions which can be the sites of cross-linking. Preferred hydrophobic
25 groups are alkyl groups of more than six , preferably eight, and even more preferably 12 carbon atoms, and derivatives thereof.

Chemical cross-linking is defined as forming a chemical bond between the cross-linked polymers. Chemical cross-linking is
30 understood to be stable and thermally irreversible. Some of the advantages in the use of hydrophobically modified polymers are responsiveness to hydrocarbons resulting in a reduction in the adhesive strength of gels and an extreme retardation of

gelation, and solubilisation of large organic cross-linking agents.

The cross-linking agents can be either inorganic ions (or ionic
5 complexes) or polar organic molecules. When the polymer contains ionic groups such as carboxylate or sulphonate functions the polymer chains can be cross-linked by inorganic ions such as chromium(III) or zirconium(IV), frequently in the presence of ligands, such as acetate or adipate ions, to control the rate of
10 cross-linking. Alternatively, organic cross-linking agents can be used. For example, the amide groups on poly(acrylamide) can be cross-linked with formaldehyde or a mixture of formaldehyde and phenol. The disadvantages of using formaldehyde or similar small molecules are their toxicity and issues of disposal. The
15 larger reagents such as hexanal and heptanal, which are not classed as toxic, would be suitable cross-linking agents but are not sufficiently soluble in water to cross-link normal hydrophilic polymers such as the poly(acrylamide)/poly(acrylate) copolymer. However, these larger cross-linking agents are
20 solubilised in aqueous solutions by hydrophobically-modified poly(acrylamide) polymers and can thence cross-link them.

These and other features of the invention, preferred embodiments and variants thereof, and further advantages of the invention
25 will become appreciated and understood by those skilled in the art from the detailed description following below.

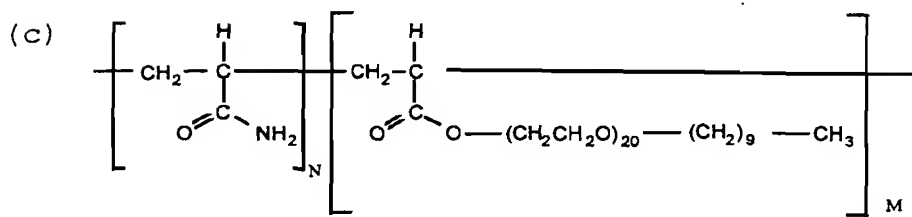
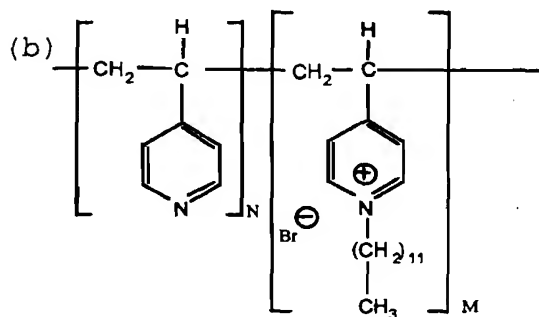
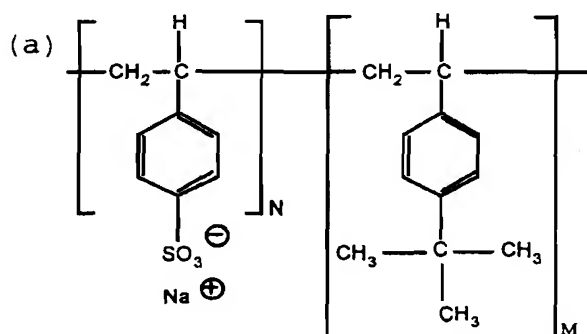
MODE(S) FOR CARRYING OUT THE INVENTION

30

A large number of hydrophobically-modified polymers are known as such and suitable for the purpose of the present invention. These polymers are based on the modification of common water-

soluble (hydrophilic) polymers such as poly(acrylamide), poly(acrylic acid), poly(vinylpyridine), hydroxyethylcellulose and poly(ethylene oxide). The structure of examples of hydrophobically-modified polymers are given below for

5 poly(sodium 4-styrenesulphonate) (a), poly(vinylpyridine) (b), and poly(acrylamide) (c):

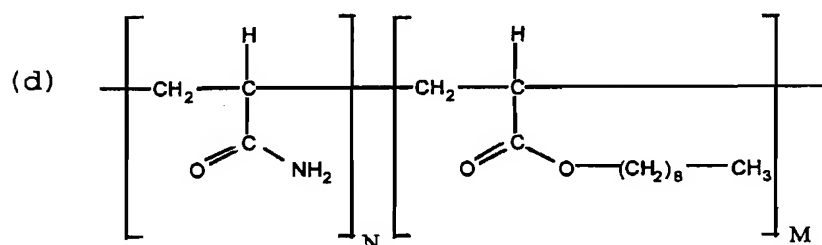


The fraction $M/(M+N)$ denotes the mole fraction of hydrophobic

15 monomers in the polymer.

The following formulae describe the structure of further examples which can be readily cross-linked using known cross-linkers.

- 5 The structure of a hydrophobically-modified poly(acrylamide) where the hydrophobic modification takes the form of replacing about 3 mole percent ($M/(M+N)=0.03$) of the acrylamide units with the hydrophobic monomer n-nonyl acrylate is:

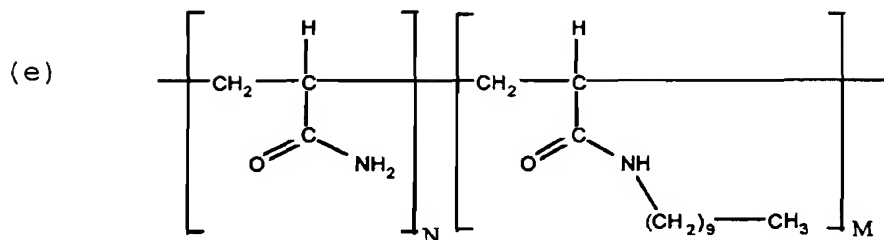


10

This polymer has an average molecular weight of about 5×10^6 g/mole and it can be readily cross-linked with chromium(III) or zirconium(IV) or the organic cross-linker hexanal. Where the gels formed by conventional poly(acrylamide)/poly(acrylate) copolymers undergo syneresis and shrink, the gels formed by hydrophobically-modified polymers in otherwise identical formulations undergo syneresis at approximately half the rate of the hydrophilic polymers. It is also observed that the cross-linking of the hydrophobically-modified poly(acrylamide) is less sensitive to the concentration of divalent ions (Ca^{2+} , Mg^{2+}).

The structure of another hydrophobically-modified poly(acrylamide) where the hydrophobic modification takes the form of replacing about 1 mole percent ($M/(M+N)=0.01$) of the acrylamide units with the hydrophobic monomer N-decylacrylamide is:

25



This polymer can also be cross-linked using metal ions (Cr(III), Zr(IV)) and organic molecules (hexanal).

5 An additional feature of hydrophobically-modified polymers such as (d) and (e) is that they can be prevented from cross-linking in contact with hydrocarbon. The prevention of cross-linking can arise from two possible mechanisms. Firstly, the larger organic
 10 cross-linker can be removed from the polymer solution by solubilisation in the contacting hydrocarbon. The cross-linker hexanal, which is solubilized by the hydrophobically-modified polyacrylamide, is readily soluble in hydrocarbon and can be extracted from the polymer solution. Secondly, the solution of
 15 hydrophobically-modified polymer can partly solubilize the hydrocarbon which itself can be sufficient to prevent cross-linking. Thus a solution of hydrophobically-modified poly(acrylamide) with the metal ion cross-linker such as Cr(III) or a small water-soluble cross-linker such as formaldehyde can
 20 be prevented from cross-linking when mixed with hydrocarbon.

CLAIMS

1. A hydrophobically modified water soluble polymer capable of being chemically cross-linked so as to produce a stable gel
5 for blocking a water-bearing formation from a hydrocarbon-producing well, characterized in that the polymer is essentially linear, thus defining a polymeric backbone, and hydrophobic groups are located at random positions along the backbone.
- 10 2. The polymer of claim 1, comprising 0.5 to 5 mole per cent of a hydrophobic monomer.
- 15 3. The polymer of claim 1, comprising functional groups capable of being chemically cross-linked, said groups being part of a hydrophilic section of the polymer.
- 20 4. The polymer of claim 1, comprising functional groups capable of being chemically cross-linked, said groups being part of a hydrophobic section of the polymer.
5. The polymer of claim 1, having a molecular weight of 50,000 or more.
- 25 6. A composition for selectively blocking water bearing formations from a hydrocarbon producing well comprising a polymer in accordance with claim 1 and a chemical cross-linking agent.
- 30 7. The composition of claim 6, wherein the chemical cross-linking agent is organic.

8. The composition of claim 6, wherein the chemical cross-linking agent is an aldehyde or aldehyde derivative comprising at least 5 carbon atoms.
- 5 9. The composition of claim 6, wherein the chemical cross-linking agent is hexanal or heptanal.



Application No: GB 9805880.3
Claims searched: 1-9

Examiner: Dr Paul R Minton
Date of search: 15 June 1998

Patents Act 1977
Search Report under Section 17

Databases searched:

UK Patent Office collections, including GB, EP, WO & US patent specifications, in:

UK Cl (Ed.P): C3J (JCC); C3P (PFE); E1F (FPC).

Int Cl (Ed.6): E21B 33/138, 43/22.

Other: ONLINE: WPI

Documents considered to be relevant:

Category	Identity of document and relevant passage	Relevant to claims
X	EP 0 399 767 A2 (HALLIBURTON). See particularly Claims 1-12.	1-6
X	EP 0 311 799 A1 (AMERICAN CYANAMID). See particularly Claims 1-11.	1-5
X	US 5 003 006 A (CHU <i>et al</i>). See particularly lines 9-14, column 4 and Claims 1-11.	1-9
X	US 4 814 096 A (EVANI). See particularly Claims 1-5.	1-3,5
X	US 4 520 182 A (TURNER <i>et al</i>). See particularly Claims 1-8.	1-3,5
X	US 4 505 827 A (ROSE <i>et al</i>). See particularly lines 35-67, column 4.	1-3,5
X	US 3 984 333 A (KRAATS <i>et al</i>). See particularly Table 1.	1-3,5

X Document indicating lack of novelty or inventive step
Y Document indicating lack of inventive step if combined with one or more other documents of same category.

& Member of the same patent family

A Document indicating technological background and/or state of the art.
P Document published on or after the declared priority date but before the filing date of this invention.

E Patent document published on or after, but with priority date earlier than, the filing date of this application.